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## Structure Reports

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## 4-Amino-6-tert-butyl-3-thioxo-3,4-dihydro-1,2,4-triazin-5(2H)-one

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## Key indicators

Single-crystal X-ray study
$T=273 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.038$
$w R$ factor $=0.115$
Data-to-parameter ratio $=14.7$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]In the title compound, $\mathrm{C}_{7} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{OS}$, the triazine ring is planar. The crystal structure is stabilized by intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds which link the molecules into chains running along the $c$ axis.

## Comment

1,2,4-Triazines and the compounds derived from them are found to possess a wide variety of pharmacological activities (Holla et al., 1999). Triazine derivatives include herbicides having a broad spectrum of action that kill many species of weeds, and also herbicides with a narrow selectivity (Gruzdyev et al., 1983). The title compound, (I), is an intermediate in the synthesis of a herbicide, 4-amino-6-(tert-butyl)-3-methylthio-1,2,4-triazin-5(4H)-one, commonly known as 'Sencor' (Eue \& Tietz, 1972). It is used for the control of weeds in potato crops and is a broad-spectrum herbicide used in pre-emergence application in potato crops at a rate of $0.75-1 \mathrm{~kg}$ per hectare. In view of the importance of the title compound, (I), its crystal structure is reported here.


In all essential details, the molecular geometry (Fig. 1 and Table 1) is in good agreement with related structures (Ghassemzadeh et al., 1998). The triazine ring is planar and exists in the thione form; the $\mathrm{C}=\mathrm{S}$ bond length of 1.663 (2) $\AA$ is slightly longer than the pure double-bond distance $(1.61 \AA$; Pauling, 1960). The $\mathrm{N}-\mathrm{N}[1.341$ (2) $\AA$ ] and $\mathrm{C}-\mathrm{N}$ [mean value 1.341 (2) $\AA]$ bond distances are intermediate between the expected single ( 1.45 and $1.47 \AA$, respectively) and double ( 1.20 and $1.27 \AA$, respectively) bond distances.

The crystal structure is stabilized by intra- and intermolecular hydrogen bonds (Table 2). Atom N1 of the triazine ring links the carbonyl O atom of an adjacent triazine ring, forming a chain running parallel to the $c$ axis. The tertiary butyl group is oriented such that there are alternate hydrophobic and hydrophilic layers in the crystal packing (Fig. 2).

## Experimental

The title compound, (I), was obtained as a gift sample from Rallies India Ltd, Bangalore. The compound was recrystallized by slow evaporation of an acetone-toluene (9:1) solution (m.p. 488 K ).

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Figure 1
A view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms are shown as small spheres of arbitrary radii.


Figure 2
A packing diagram of (I), viewed down the $b$ axis, showing the chain formation. Dashed lines indicate $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds. H atoms not involved in hydrogen bonding have been omitted for clarity.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{7} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{OS} \\
& M_{r}=200.27 \\
& \text { Monoclinic, } P 2_{1} / c \\
& a=12.5196(8) \AA \\
& b=6.9535(4) \AA \\
& c=12.5987(8) \AA \\
& \beta=112.039(1)^{\circ} \\
& V=1016.64(11) \AA^{3} \\
& Z=4
\end{aligned}
$$

## Data collection

Bruker SMART APEX CCD areadetector diffractometer $\omega$ scan
Absorption correction: none 9276 measured reflections 1780 independent reflections
$D_{x}=1.308 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 6656 reflections
$\theta=3.0-27.9^{\circ}$
$\mu=0.29 \mathrm{~mm}^{-1}$
$T=273$ (2) K
Block, colorless
$0.21 \times 0.17 \times 0.09 \mathrm{~mm}$

1640 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.018$
$\theta_{\text {max }}=25.0^{\circ}$
$h=-14 \rightarrow 14$
$k=-8 \rightarrow 8$
$l=-14 \rightarrow 14$

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0662 P)^{2}\right. \\
& \quad+0.2933 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.40 \mathrm{e}^{2} \AA^{-3} \\
& \Delta \rho_{\min }=-0.35 \mathrm{e}^{-3}
\end{aligned}
$$

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.039$
$w R\left(F^{2}\right)=0.115$
$S=1.07$
1780 reflections
121 parameters
H -atom parameters constrained

Table 1
Selected geometric parameters ( $\mathrm{A},{ }^{\circ}$ ).

| C1-N1 | $1.336(3)$ | $\mathrm{C} 2-\mathrm{N} 2$ | $1.378(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C} 1-\mathrm{N} 2$ | $1.370(2)$ | $\mathrm{C} 3-\mathrm{N} 3$ | $1.288(2)$ |
| $\mathrm{C} 2-\mathrm{O} 1$ | $1.2244(19)$ | $\mathrm{N} 2-\mathrm{N} 4$ | $1.401(2)$ |
|  |  |  |  |
| $\mathrm{C} 1-\mathrm{N} 1-\mathrm{N} 3$ | $127.70(15)$ |  |  |

Table 2
Hydrogen-bond geometry ( $\mathrm{A},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| N1-H1 $\cdots \mathrm{O}^{\text {i }}$ | 0.86 | 1.94 | $2.790(2)$ | 168 |
| N4-H4B $\cdots$ O1 | 0.86 | 2.31 | $2.607(2)$ | 100 |
| N4-H4A $\cdots$ S1 | 0.86 | 2.59 | $2.976(2)$ | 108 |

Symmetry code: (i) $x,-y+\frac{1}{2}, z-\frac{1}{2}$.
All H atoms were included in calculated positions ( $\mathrm{C}-\mathrm{H}=0.93-$ $0.98 \AA$ and $\mathrm{N}-\mathrm{H}=0.86 \AA$ ) and refined using a riding model, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{N}, C)$ or $1.5 U_{\text {eq }}$ (methyl C). The methyl groups were allowed to rotate but not to tip.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL/PC (Sheldrick, 1990) and MERCURY (Bruno et al., 2002); software used to prepare material for publication: SHELXL97.

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